

# MICROBEADS - A PROMISING TECHNOLOGY FOR CONTROLLED DRUG DELIVERY: A REVIEW

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#### **Abstract:**

Microbeads have emerged as a versatile and effective platform for controlled drug delivery, offering advantages such as improved bioavailability, reduced dosing frequency, and enhanced therapeutic efficacy. This review provides a comprehensive overview of recent advancements in the formulation and evaluation of microbeads for pharmaceutical applications. We discuss various polymers utilized in microbeads preparation, including natural (e.g., alginate, chitosan) and synthetic (e.g., poly (lactic-co-glycolic acid), Eudragit) materials, and their impact on drug release kinetics. The article examines different techniques for microbeads fabrication, such as Ionotropic gelation, emulsification, and spray drying, highlighting their principles, advantages, and limitations. Additionally, we explore strategies for optimizing drug loading, encapsulation efficiency, and release profiles. The review also covers state-of-the-art characterization methods for evaluating microbeads properties, including size distribution, surface morphology, zeta potential, and in vitro drug release studies. Furthermore, we address the challenges in microbeads formulation and potential solutions, as well as regulatory considerations for their development. Finally, we provide insights into emerging trends and future directions in microbeads technology for drug delivery. This comprehensive review aims to serve as a valuable resource for researchers and formulators working on microbeads-based drug delivery systems.

**Keywords:** microbeads, controlled release, drug delivery, polymer-based carriers.

## **Introduction:**

The thing in designing delayed release sustained or controlled delivery system is to reduce the frequency of dosing or to increase the effectiveness of the medicine by localization at the point of action, reducing the cure needed, or furnishing invariant medicine delivery. Sustained- release, sustained action, dragged action, extended action are the terms used to identify medicine delivery systems that are designed to achieve a prolonged remedial effect by continuously releasing drug over an extended period of time after administration of a single cure. The design of effective medicine delivery systems has lately come an integral part of the development of new drugs.

Hence, exploration continuously keeps on searching for ways to deliver medicines over an extended period of time, with a well- controlled release profile. (1)

Thus, suitable synthetic styles for generation of bio-comparable glamorous globules have attracted significant interest in the recent times. (2)

Microbeads have a periphery of 0.5- 1000 µm and are nearly globular in shape. Treatment with different active agents can be carried out with multitudinous release biographies or a sustained release with minimum adverse goods thanks to the solid and free- flowing particulate carriers that contain dispersed medicine patches in crystalline or result form. Likewise, under physiological settings, the microbeads remain effective. They can also be modified to include specifics and deliver them locally at high attention, ensuring that remedial quantities are achieved at the target point and minimizing negative goods by maintaining low systemic attention. A variety of polymers, including cationic polymers like chitosan, anionic polymers like sodium alginate, and binding factors like gelatin, chondroitin sulphate, and avidin, are combined in a preset rate to produce the Microbeads. (3, 4)

Microbeads are characterized as free flowing multi particulate system which consists of proteins or synthetic polymers which are biodegradable in nature and immaculately having a flyspeck size lower than 200 μm.(5)

The design of effective medicine delivery systems has lately come an integral part of the development of new drugs. Hence, exploration continuously keeps on searching for ways to deliver medicines over an extended period of time, with a well- controlled release profile. Oral medicine delivery is the most desirable and favored system of administering remedial agents for their systemic goods. In addition, the oral drug is generally considered as the first avenue delved in the discovery and acceptance, convenience, and cost effective development of new medicine realities and pharmaceutical phrasings, substantially because of case manufacturing process.(6)

For numerous decades treatments of an acute complaint or a habitual illness has been substantially fulfilled by delivery of medicines to cases using colorful conventional pharmaceutical lozenge like tablets, capsules, capsules, suppositories, creams, ointments, liquids, aerosols and Injectable as medicine carriers. This type of medicine delivery system is known to give a prompt release of medicine.

So to achieve and maintain the medicine attention within therapeutically effective range demanded for treatment, it's frequently essential to take this type of medicine delivery system several times a day which results in a significant change in medicine situations. For numerous medicine substances, conventional immediate release phrasings give clinically effective remedy while maintaining the needed balance of pharmacokinetic and pharmacodynamic biographies with respectable position of safety to the case (7).

## **MICROBEADS:**

Microbeads, as the name suggests they're nearly globular, small with periphery of 0.5- 1000μm in size, solid and free flowing particulate carriers containing dispersed medicine patches either in result or crystalline form that allow a sustained release or multiple release biographies of treatment with colorful active agents without major side goods.

Also, the globules maintain functionality under physiological conditions, can incorporate medicine to deliver locally at high attention icing that remedial situations are reached at the target point while reducing the side goods by keeping systemic attention low.

The microbeads are produced from several polymers similar as cationic polymers e.g. chitosan, anionic polymers e.g. sodium alginate, and binding factors e.g. gelatin, chondroitin sulfate, avidin in destined rate. (8-9)

## **METHODS OF PREPARATION: (10)**

Preparation of microbeads should satisfy certain criteria

- 1. Stability of the medication after conflation with a clinically respectable shelf life.
- 2. Controlled flyspeck size and dispersability in waterless vehicles for injection.
- 3. The capability to incorporate nicely high attention of the medicine.
- 4. Release of active reagent with a perfect control over a wide duration.
- 5. Biocompatibility with a controllable biodegradability and vulnerability to chemical revision.

# Biopharmaceutical classification system;

The Biopharmaceutical Bracket System (BCS) involves placing a medicine into four groups:

- High solubility and high permeability
- Low solubility and high permeability
- High solubility and low permeability
- Low solubility and low permeability
- Class 1 is considered the favored order, while Class 4 is the worst order. A medicine having high solubility in the intestine is a good medicine for a controlled oral lozenge form. The medicine permeability value must also be considered and should be further than the prescribed value.

A natural half- life of a needed medicine is between two and six hours is the stylish choice of expression because this type of criteria of the medicine is avoiding the accumulation of the medicine in the body. (11-13)

# Application;

- • In order to give exfoliation, microbeads are added to toothpaste, face diminutives, detergents, and other cosmetics and particular hygiene particulars. To make over-the-counter specifics simpler to swallow, they could be added.
- Microbeads are utilized in fluid visualization, process troubleshooting, microscopy techniques, and fluid flow analysis in biological and health science research(20)
- Creams and lotions have a smooth texture and are easily spread because of the ball-bearing effect caused by sphericity and uniform particle size.
- Roundness and smoothness can act as lubricants. Cosmetic goods look more appealing when they contain colored microspheres(14)

# Preformulation Studies :( 15)

# Organoleptic properties of drug:

The Organoleptic properties like physical state, colour, taste, odour, etc., of the drug were reported with help of the descriptive terminology. It helps to identify the drug.

## **Melting point:**

In this method a small amount of drug was filled in a capillary tube of open both the ends and it was placed along with thermometer in melting point apparatus.

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# **Solubility profile:**

It is important to know about solubility characteristics of a drug in aqueous system, since they must possess some limited aqueous solubility to elicit a therapeutic response.

Determination of Percentage yield the % yield of all the formulations of alginate microbeads are calculated using the formula.

Determination of Bulk Density (Db) it is the ratio of total mass of powder to the bulk volume of powder. It is measured by putting the alginate microbeads into a measuring cylinder and initial capacity is noted. This initial capacity is said to be bulk volume. From this the bulk density is calculated using the formula mentioned below, it is expressed in gm/ml and is given by

$$Db = M / Vb$$

Where, M - Mass, Vb - bulk volume, Db - bulk density.

# **Determination of Tapped Density (Dt):**

Tapped density is determined by placing a graduated cylinder, containing a known mass of microbeads. The cylinder is allowed to fall under its own weight onto a hard surface from the height of 10 cm at 2 seconds intervals.

## **Determination of Hausner ratio:**

It is an another parameter for measuring the flowability of prepared alginate micro beads and is calculated using the formula,

#### **Determination of Particle size:**

Determination of flyspeck size for size distribution analysis was determined using an optic microscopy system. Roughly 100 micro patches were counted for flyspeck size using a calibrated optic microscope.

## **Techniques of Microbeads**

# **Ionotropic Gelation Method:**

To begin cross-linking, an ionic polymer interacting with an oppositely charged ion is all that is required. The electroneutrality principle cannot fully account for the interaction of polyanion with cations, in contrast to simple

monomeric ions. The capacity of cations to conjugate with anionic functions or vice versa is influenced by the three-dimensional structure and the presence of other groups (16)

#### **External Gelation Method:**

As a source of the cross-linking ion in the external gelation process, a essence ion result is employed. A needle is used to gently stir the medicine- containing polymer result before it's extruded into the admixture. Tone-sustaining blob conformation is the result of immediate gelation that takes place as soon as the polymeric drop comes into contact with the essence ion result. Before being taken out and dried, the globules are cured for a destined quantum of time in the gelation medium. The cross-linker ions verbose snappily into the incompletely gelated globules, causing the external gelation to be (17)

#### **Internal Gelation Method:**

In the internal gelation process, the cross-linker ion is created "in situ." By dwindling the pH of the result, the cation is liberated in situ along with the essence ion and the essence swab (18)

## **Emulsion Gelation Method:**

Emulsion gelation procedures are another way to prepare microbeads. By dispersing the counted quantum of sodium alginate in deionized water, the sodium alginate result was created. To gain a homogenous medicine-polymeric admixture, a precisely counted volume of medicine was introduced to a polymeric result of sodium alginate and the medicine was magnetically agitated with low heat. A certain quantum of cross-linking agent was added to produce a thick dissipation, which was also extruded into oil painting containing span 80 and 0.2 glacial acetic acid using hype fitted with a flat- sloped needle of size no. 23 while being stirred magnetically at 1500 rpm. To produce stiff, distinct patches, the microbeads are left in the oil painting for thirty twinkles. They were collected by decantation and the products therefore separated was washed with chloroform to remove the traces of oil painting the microbeads were dried at 400°C for 12 h (19)

# **Polyelectrolyte Complexation Method:**

A fresh fashion for creating microbeads is the complex coacervation of polyelectrolytes with opposing charges, polycation and polyanion accourrements, and alginate- chitosan microcapsules that are biocompatible and biodegradable. These microcapsules can be produced in mild or indeed physiological conditions, making them applicable for use in biomedical fields.

The use of alginate – chitosan microcapsules as medicine- delivery vehicles for proteins and polypeptides has drawn further attention in recent times. Using this fashion, the admixture will resolve into a dilute equilibrium phase and a thick coacertive phase that contains the microbeads, depending on the pH, ionic strength, and polyion attention.

By scattering the sodium alginate result into the chitosan result, for case, complicated coacervation between alginic acid and chitosan was fulfilled, performing in robust microbeads that remained stable over a wide pH range. The optimal yield when using coacervative globules requires medication conditions of pH 3.9, ionic strength of 1 mM, and total polyion content of 0.15 w/v. (20)

## **Double emulsion technique;**

Double conflation system of microbeads medication involves the conformation of the multiple mixes or the double conflation of type w o/ w and is stylish suited to water answerable medicines, peptides, proteins and the vaccines.

This system can be used with both the natural as well as synthetic polymers. The waterless protein result is dispersed in a lipophilic organic nonstop phase.

This protein result may contain the active ingredients. The nonstop phase is generally comported of the polymer result that ultimately encapsulates of the protein contained in dispersed waterless phase. The primary conflation is subordinated also to the homogenization or the sonication before addition to the waterless result of the poly vinyl alcohol (PVA).

This results in the conformation of a double conflation. The conflation is also subordinated to solvent junking either by solvent evaporation or by solvent birth. a number of hydrophilic medicines like luteinizing hormone releasing hormone (LH- RH) agonist, 391 vaccines, proteins/ peptides and conventional motes are successfully incorporated into the microbeads using the system of double conflation detergent evaporation/ extraction (21).

# Spray drying and spray congealing

These styles are grounded on the drying of the mist of the polymer and medicine in the air. Depending upon the junking of the detergent or cooling of the result, the two processes are named spray drying and spot congealing independently. The polymer is first dissolved in a suitable unpredictable organic detergent similar as dichloromethane, acetone, etc. The medicine in the solid form is also dispersed in the polymer result under high speed homogenization. This dissipation is also comminuted in a sluice of hot air. The atomization leads to the conformation of the small driblets or the fine mist from which the solvent evaporates presently leading the conformation of the microspheres in a size range 1- 100 µm.

Microbeads are separated from the hot air by means of the cyclone separation while the traces of detergent are discarded by vacuum drying. One of the most advantages of the process is feasibility of operation under aseptic conditions.

The spray drying process is used to synopsize colorful penicillin. Thiamine mononitrate 14 and sulphaethylthiadizole 15 are reprised in an admixture of mono- and diglycerides of stearic acid and palmitic acid using spray congealing. Veritably rapid-fire detergent evaporation, still leads to the conformation of pervious microbeads (22)

## **Solvent extraction**

Solvent evaporation method is used for the medication of microbeads, involves junking of the organic phase by extraction of the organic detergent. The system involves water miscible organic detergents similar as isopropanol. Organic phase is removed by birth with water. This process decreases the hardening time for the microbeads. One variation of the process involves direct addition of the medicine or protein to polymer organic result. The rate of solvent junking by birth system depends on the temperature of water, rate of conflation volume to the water and the solubility profile of the polymer (23).

## Phase separation coacervation technique

This process is grounded on the principle of dwindling the solubility of the polymer in organic phase to affect the conformation of polymer rich phase called the coacervates. In this system, the medicine patches are dispersed in a result of the polymer and an inharmonious polymer is added to the system which makes first polymer to phase separate and gulf the medicine patches. Addition of non-solvent results in the solidification of polymer. Poly lactic acid (PLA) microbeads have been prepared by this system by using butadiene as inharmonious polymer.

The process variables are veritably important since the rate of achieving the coacervates determines the distribution of the polymer film, the flyspeck size and agglomeration of the formed patches. The agglomeration must be avoided by stirring the suspense using a suitable speed stirrer since as the process of microbeads conformation begins the formed polymerize droplets start to stick and form the agglomerates, thus the process variables are critical as they control the kinetic of the formed patches since there's no defined state of equilibrium attainment (24).

# **Evaluation of Microbeads [27-31]**

# 1) Size and shape of microbeads

Light microscopy (LM) and scanning electron microscopy (SEM) each may can be used to decide the size, shape and outer structure of microbeads

# 2) Drug content and Entrapment efficiency

Accurately weighed microbeads - 100mg have been suspended in 100ml of simulated intestinal fluid of pH 6.8 and keep for 24hrs. Next day it was stirred for 5min and filtered through the use of Whatman filter paper. After appropriate dissolution, the drug content material within side the filtrates were analyzed spectrophotometrically at 241nm the use of UV spectrophotometer.

Drug entrapment efficiency may be calculated using following equation,

% Entrapment = Actual content material/Theoretical content material x 100

### 3) Swelling properties

The swelling was measured by % weight gain in the beads. The swelling behaviors of all the formulations were studied. In this test 20 mg of beads from each formulation was kept in Petri dish containing distilled water. At the final of one hour, the beads were withdrawn, soaked with tissue paper and weighed.

Then for every one hour interval, weights of beads were recorded and the process was last till the end of 8 hours.

The mucoadhesive microbeads to get swelled at the absorbing surface by absorbing fluids at the site of absorption, which is a primary requirement for initiation of mucoadhesion. The swelling index of the microbeads was calculated by using the formula,

Swelling index = (mass of swollen microbeads – mass of dry microbeads/mass of dried microbeads) 100

(Or)

% of swelling = DT - D0 / D0  $\times$  100

Where,

D0 = Weight of dried microbeads

DT = Weight of swelled microbeads

## 4) Floating time and lag floating time

Specified weight (50 mg) of floated beads was placed in a beaker containing 100 ml of buffer 1.2 pH and shake at 50 rpm in a water bath  $37 \pm 0.5$ °C.

The duration took by beads to float on the surface was analysed (lag time). Buoyant duration of time was calculated by the following equation:

(%) floating = weight of floated beads (Wf)/weight of floated beads (Wf) + weight of settled beads (Ws)\*100

# 5) FTIR of microbeads

FTIR spectral measurement was performed using FTIR spectrophotometer to find the presence of any interaction between the polymer and drug.

The polymer and the microbeads were finely ground with KBr to produce the pellets under a hydraulic pressure of 600psi and spectra were scanned between 400 and 4000cm<sup>-1</sup>.

## **6) SEM**

The drug-loaded beads obtained from various percentages of polymer, CaCl2 and drug were studied by using a scanning electron microscope.

The beads were coated with carbon and gold (100 and 50 Å thickness respectively) in a vacuum evaporator in an argon atmosphere.

The coated substances were then analysed under a scanning electron microscope operated at 15 KV.

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